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The purpose of the lab is to experimentally calculate the concentration of NaOH using a titration with 10 mL of 1.5M HCl. Procedure. 1. Add 50 mL of an unknown concentration of NaOH to the buret. ... Answer: "H" if it would have caused your calculated value for Molarity of NaOH to come out too high

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### **Liquid Chromatography**

There are two parts to this lab. In part one, the two dyes in the drink are separated using 10 mL of 70% isopropanol as the eluant. One mL of Kool-Aid is injected into the chromatography column and then a constant supply of the eluant is applied so that the dyes are flushed through the column at different rates. Measurements are recorded of the

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